Production of biofuel's precursors from HTL of industrial residues

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Heat-to-Fuel interfaces to advanced Power-to-Gas and Power-to-Liquids Technologies (e-fuels) workshop

Fuel production session





Hydrothermal liquefaction Feedstock in Heat to Fuel Batch experiments

- Experimental setup
- Biocrude yield
- Biocrude characterization
- Reaction mechanism

Continuous experiments

- Experimental setup
- Main challenges
- Biocrude characterization

Conclusion

Hydrothermal liquefaction



- Thermochemical process occurring in sub/supercritical water
- Mimics crude oil formation (high pressure and temperature) in minutes instead of thousands of years
- Reaction medium is water \Box indicated to treat wet biomass (algae, sewage sludge, lignin-stream, etc.) 400



Feedstock in HtF

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Project focus on lignin-rich cake from cellulosic ethanol



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BATCH EXPERIMENTS



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Experimental setup



Custom tubular reactor in hot fluidized sand bed

- Biomass-water ratio: 10-20 wt% (d.b.)
- Reaction temperatures: 300 370°C
- Residence time: 5-20 min
- Heating rate: 37 41°C min⁻¹



BV: ball valve; NV: needle valve; PR: pressure regulator; PRV: pressure relief valve

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- Two-step solvent extraction:
 - Light and heavy biocrude fraction







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- •Total biocrude yield from 44.1 to 65.7 wt% (d.b.)
- •Light fraction (BC1) increased with severity
- Heavy fraction decreased
- Temperature depolymerization enhanced by KOH:
 - ↑ Relative abundance of lighter fraction
 - \downarrow Relative amount of heavier fraction



Chromatogram of a typical light biocrude

- Mainly oxygenated aromatics
- Only 20 wt% of the light fraction can be analyzed via GC-MS, the remaining fraction is constituted by lignin oligomers



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Biocrude characterization



Molecular weight

Confirmed temperature depolymerization effect



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Reaction mechanism

Influence of Temperature on monomers yield in **biocrude** and **aqueous phase**

Main influencing parameter

- Increase of aromatic monomers
- Depolymerization confirmed by methanol increase via

demethylation/demethoxylation

of the methoxyaromatic compounds

• Cracking of acids (lactic)

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High reactivity of carbonyls and cyclopentenones (c)







CONTINUOUS EXPERIMENTS



Experimental setup

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Continuous pilot unit designed and operated within the Heat to Fuel project



Experimental campaign



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Continuous experiments with lignin from cellulosic ethanol

- with Na₂CO₃ as alkaline additive
- with NaOH as alkaline additive
- without additive
 - Better performances with NaOH in terms of plant operability





Main challenges in continuous operation



□ Presence of solids (biomass) in the process stream:

- Equipment with small orifices (e.g. pump and valves) must be avoided
- Risk of settling
 Plugging. A suitable velocity must be maintained
- Alkaline additives (Na or K carbonates or hydroxides) can help to form a stable slurry with lignocellulosic feedstock
- High pressure (> 200 bar) and moderately high temperature (> 300°C) and presence of chemicals (biocrude, water-soluble organics, alkaline solution):
 - Special materials for tubing and gaskets
- ☐ Lignin fragments polymerize during heating:
 - Rapid heating rate is required to avoid blockage
- □ Biocrude (from lignin) is solid below 70-80°C:
 - The pipeline after the cooler must be heat-traced
- These issues are worsened in lab-scale plants due to the small size



Main challenges in continuous operation



Issues

- Pressure and flow rate oscillation
- Blockage in reactor, cooling and pressure let-down system
- Low residence time and heating rate



- Flow rate-driven PID control of backpressure valve
- Improving temperature management
- Stepwise heating

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- Relatively low water content
- Low ash
- Good higher heating value
- High O₂ content
- High molecular weight
- Solid at ambient temperature



Parameter	Continuous	U.M.
Water content	1.2	wt% w.b.
Ash	0.1	wt% d.b.
HHV	30.3	MJ/kg d.b.
С	69.3	wt% d.b.
Н	6.8	wt% d.b.
Ν	0.9	wt% d.b.
S	0.2	wt% d.b.
0	21.6	wt% d.b.
Mw	900	g mol⁻¹



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- Hydrothermal liquefaction is a technology indicated to process wet biomass
- Lignin cake from cellulosic ethanol produces a high molecular weight biocrude rich in aromatics
- Harsh process conditions and presence of dispersed solids challenge lab-scale continuous operation
- Further hydrotreatment or co-refining is needed to obtain a transport fuel